



**FIGURE 4.8.** Schematic diagram of pathways for the synthesis of some key molecules required for the origin of life. Formaldehyde ( $\text{CH}_2\text{O}$ ) can polymerize to produce various sugars (i.e., the formose reaction; Fig. 4.7). This polymerization is aided by the availability of reactive groups like  $\text{Ca(OH)}_2$ . Methane ( $\text{CH}_4$ ), ammonia ( $\text{NH}_3$ ), and water ( $\text{H}_2\text{O}$ ) if mixed in the presence of electrical discharges (in Miller–Urey-like settings) can lead to the production of amino acids (and other compounds). Hydrogen cyanide ( $\text{HCN}$ ) when in the presence of aqueous ammonia can produce adenine. Finally cytosine can be produced from cyanoacetylene when mixed with urea.

4.8, redrawn from Orgel L.E., *Trends Biochem. Sci.* **23**: 491–495, © 1998 Elsevier